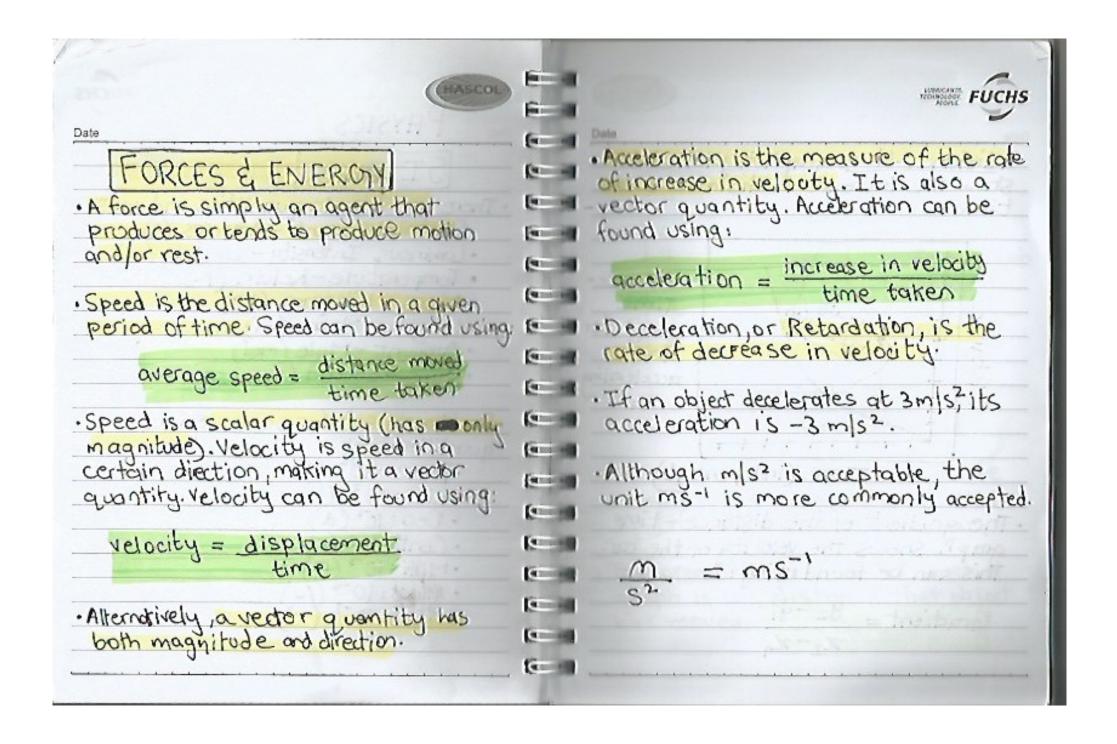
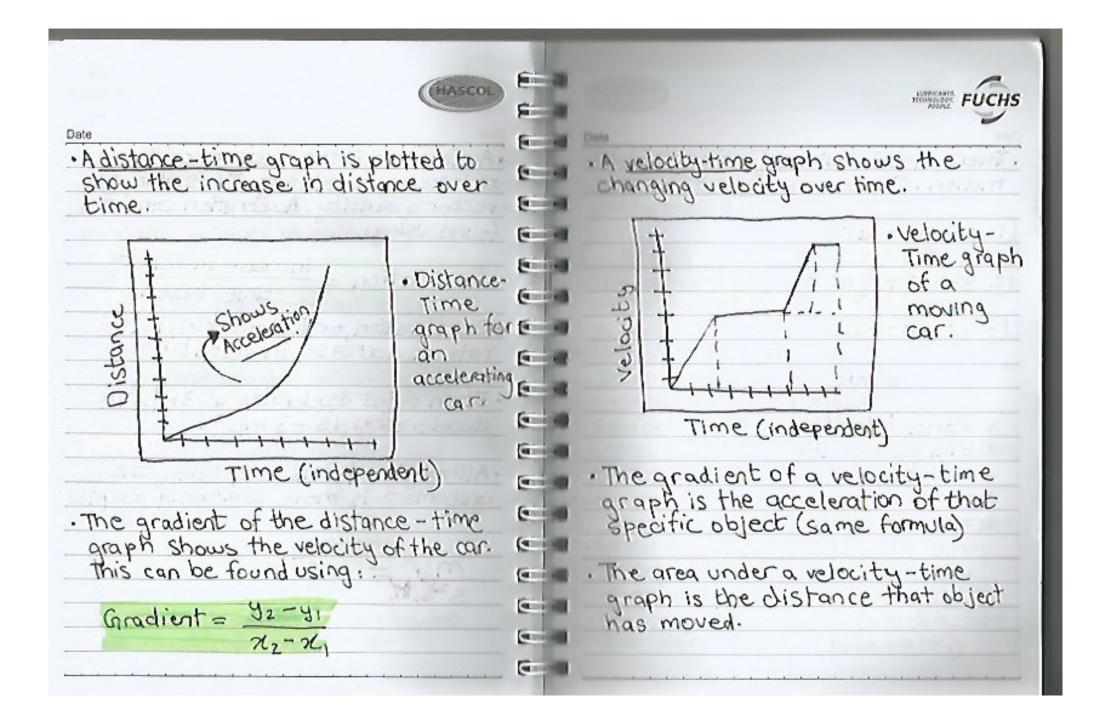
PHYSICS Date 2 SI Units 10 . There are 7 basic quantities and units: · Electrical Current - Amperes (A) · Luminous Intensity - Candela (cd) 10 · Temperature - Kelvin. (K) -· Mass - Kilogram (Kg) · Length - Metre (m) · Amount - Mole (mol) · Time - Second (s) · Prefixes used for scalar & vector quantities: · kilo : 103 (K · Deci: 10' (d · Centi: 10-2 (c) · Milli : 103 (m) · Micro: 10-6 (N · Nano: 10-1 (r





-	-		E
1AS	CO		C
		/	-
9		-	F

ALC: NO

ALC: NO

(\* · · · · ·

1

. .

( m

( ·

Concession in which the

**CONTRACTOR** 

0.000

Concession in which the

6 ....

Date

. There are 3 main equations of motion. These are:

1 V= U+ at

2 S=ut + 2at2

3 v2 = v2 + 2as

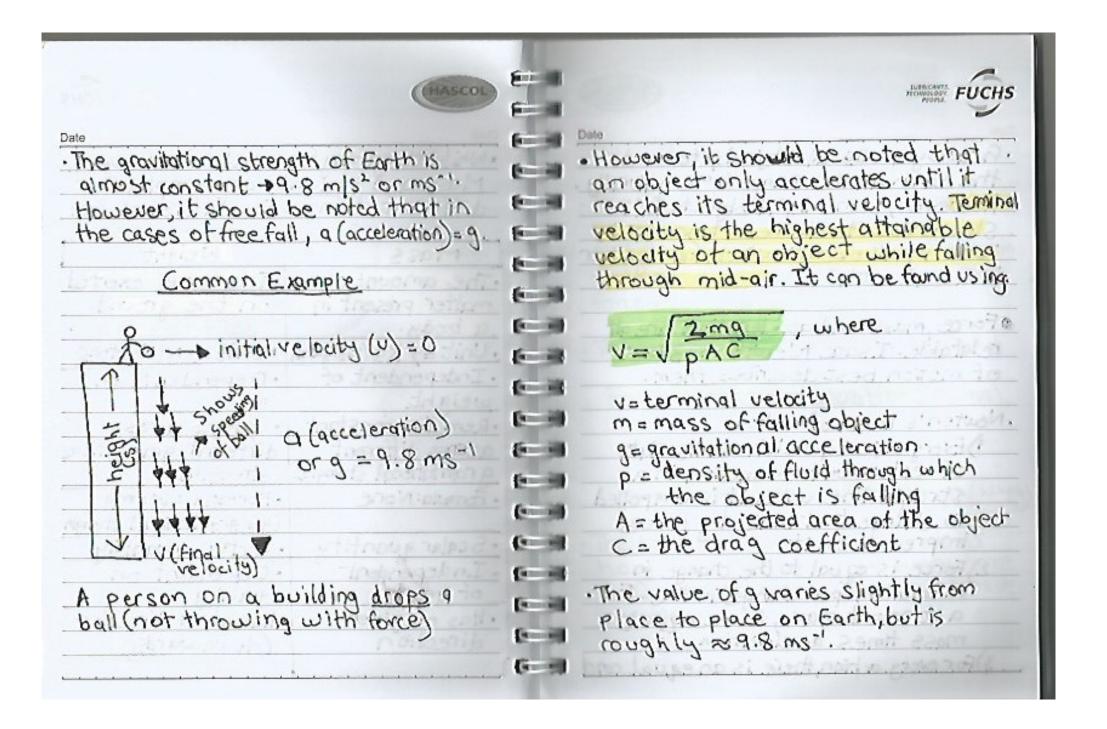
where

.....

v=final velocity U=initial velocity s=distance t=time a=acceleration ·While they may be used interchangely, Mass and Weight, are completely different things. The differences are:

FUCHS

Mass	Weight
. The amount of	. The force exerted
matter present in	on the ground.
a body.	7
· Unit => Kg (kilograms)	· Unit => N(Newtons)
. Independent of	·Dependent on mass
Remains constant	· Changes across
accoss different gravitational strengths.	different gravitations strengths.
· Formula: None	· Formula : Wamg
	(g = gravitational strength
· Scalar quantity	. Vector quantity
· Independent- of gravity	· Dependent on gravity
·Has no given direction	·Has given direction (downwards)
1	



Date

·Gravity pulls all objects towards the earth at the same time naturally. However, some light objects fall slowerdue to air resistance. This is why all objects would fall together in a vacuum.

Force, mass and acceleration are all relatable. Isaac Newton's 3 laws of motion best describes them.

Newton's Lawst of motion:
DEvery object persists in its state of rest or uniform motion in a straight line unless it is compelled to change that state by forces impressed on it.
2) Force is equal to the change, in momentum per change in time. For a constant mass, force equals mass times acceleration.
3) For every action, there is an equal and opposite reaction.

of the local division in which the

ALC: 10

C -1

. .

6 -

1.00

-

ALC: NO

.

-

-----

• The first law of motion explains how everything has inertia to change in State of motion or rest. It will remain stationary or moving unless external forces act upon it.

. The second law can be summarized:

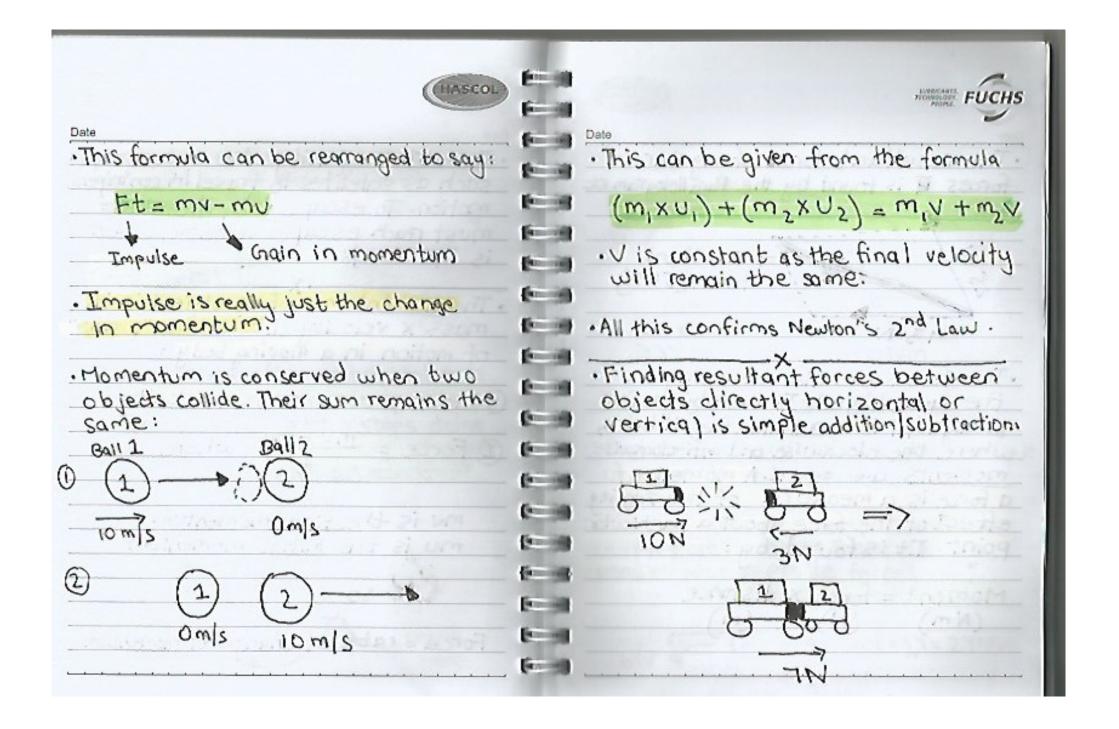
Force = mass x acceleration (F=ma),

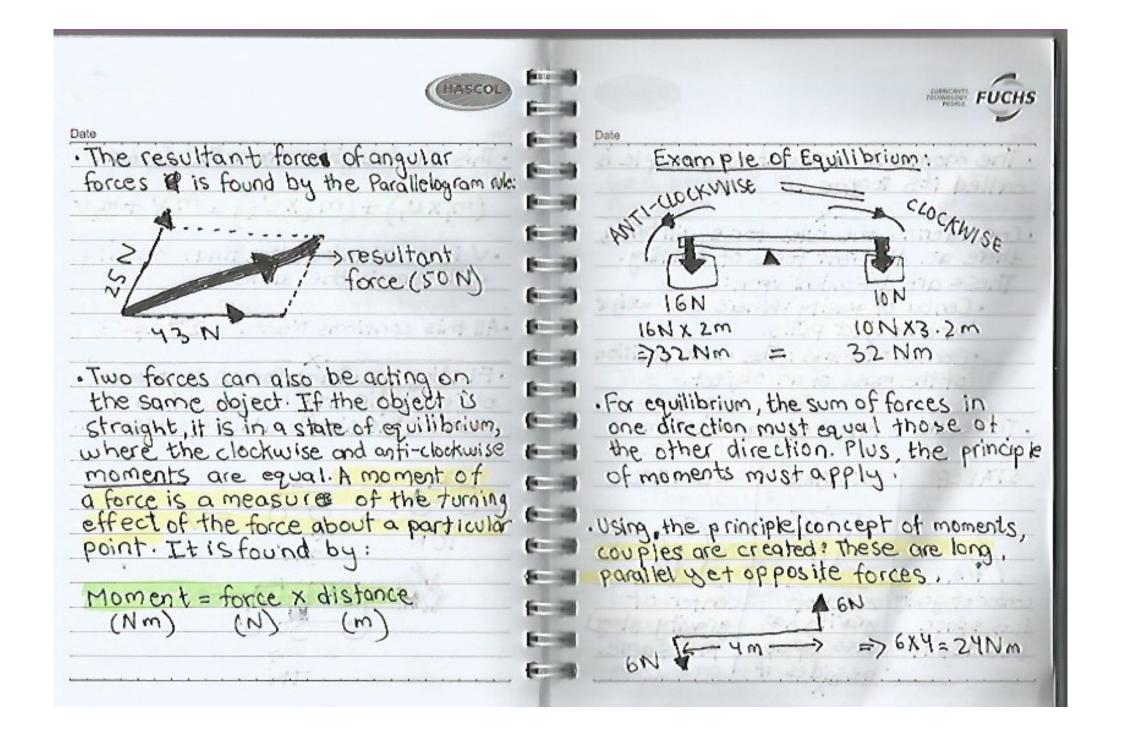
or can be replaced by

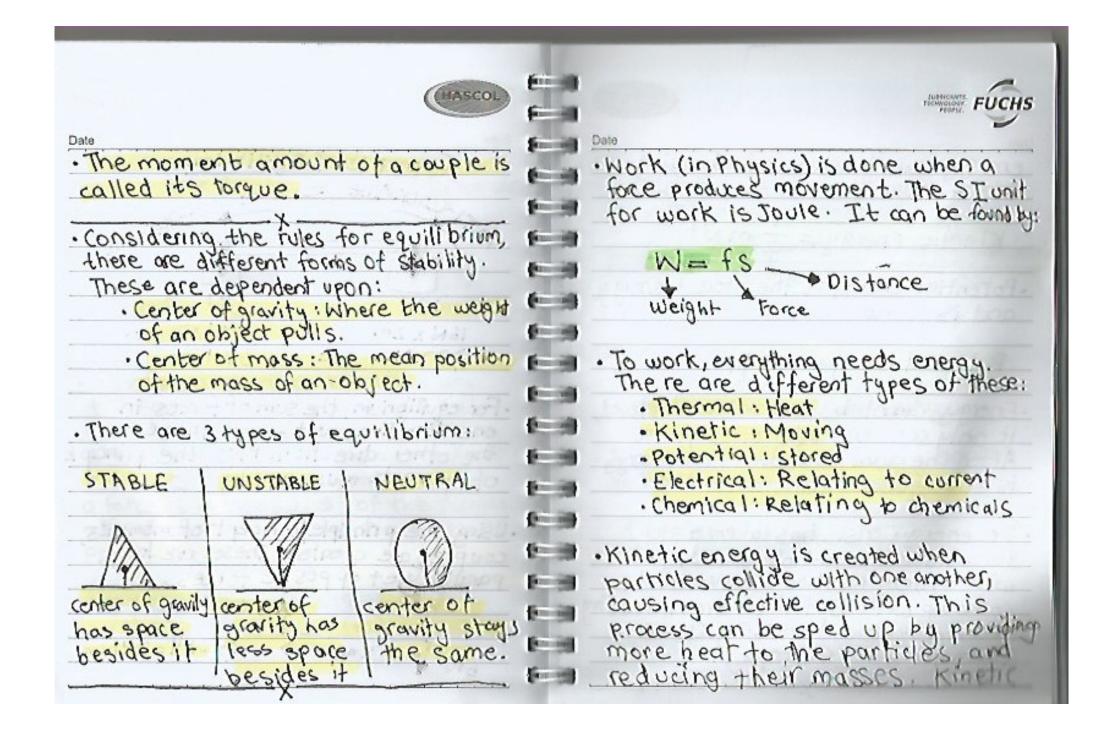
Weight = mass x gravitational strength (w=mg)

. The third law of motion explains how the net result of every force is 0, due to an equal but opposite reaction.

ASCO. Date . When materials try to slide across each other, a force called friction stops How force weight changes from Earth to moon (f=w, g=a)them. The types of friction are: · Static : The friction between Earth Moon objects that start to slide. . Dynamic: The friction during the 0 sliding m=19 m=1kg 0 . Fluid: The triction caused when 0 9=9.8 9-1.6 an object tries to slide inside agas or a liquid. . There are several types of motion: W=1.6N W=9.8N · Straight - line motion Mass is constant, while weight · Projectile motion · Contripetal changes motion Clike yene? (moving in circles) ALC: UNK ACCOUNTS ON 4. ....









6 2

1.000

Date

energy can be determined using the formula:

Kinetic Energy = - m V2

· Potential energy is the stored energy, and is found by:

Potential Energy = mgh

. Energy cannot be created or destroyed, it only converts into other forms. After the work is done, the energy is subsequently transferred.

. The energy crisis has taken a step . .... forward in the whole world, for which, new methods have to be come up with. But they are releasing carbon dioxide and causing climate ALC: NO Change (see IDU topic).

FUCHS The power of any energy source is the rate at which it gets work done. It can be measured by : Power = workdone = energy transformed time **E** · It is measured by watts (w) . In motion examples, it is true that: power = force x velocity . The efficiency of an energy source is hence also measured by: Efficiency = power output



Million and

6

(\*\*\*\*\*\*\*

10.000

STATISTICS.

411 100

121100

ALC: NO

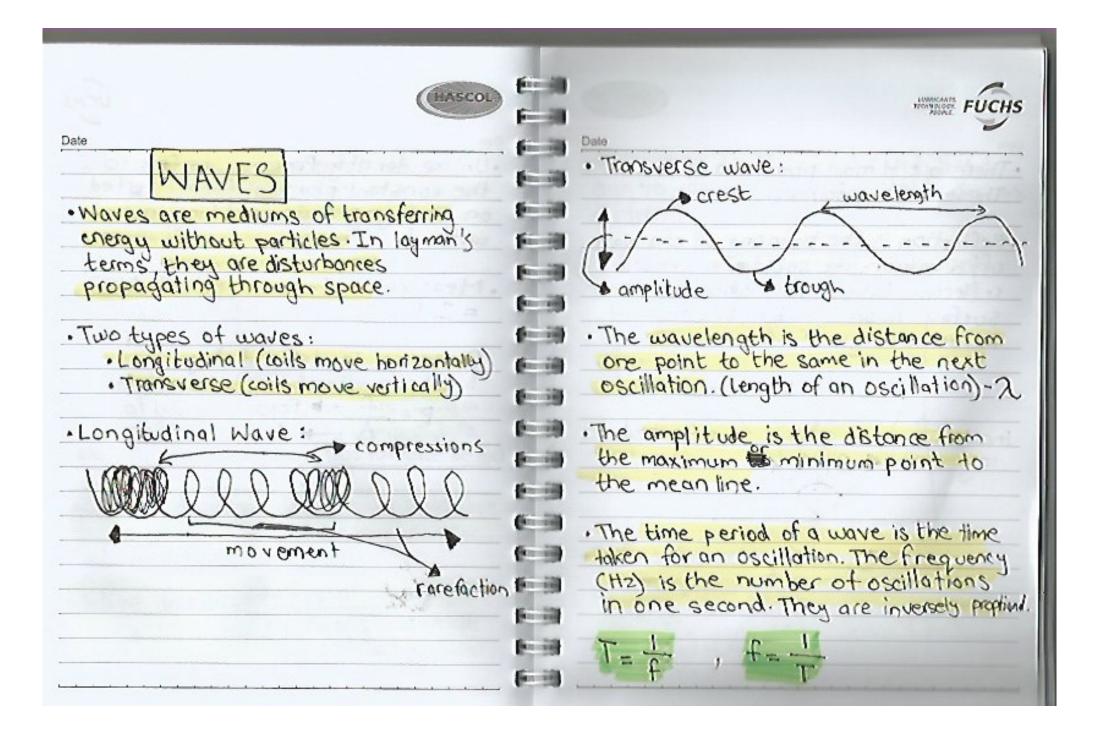
Date

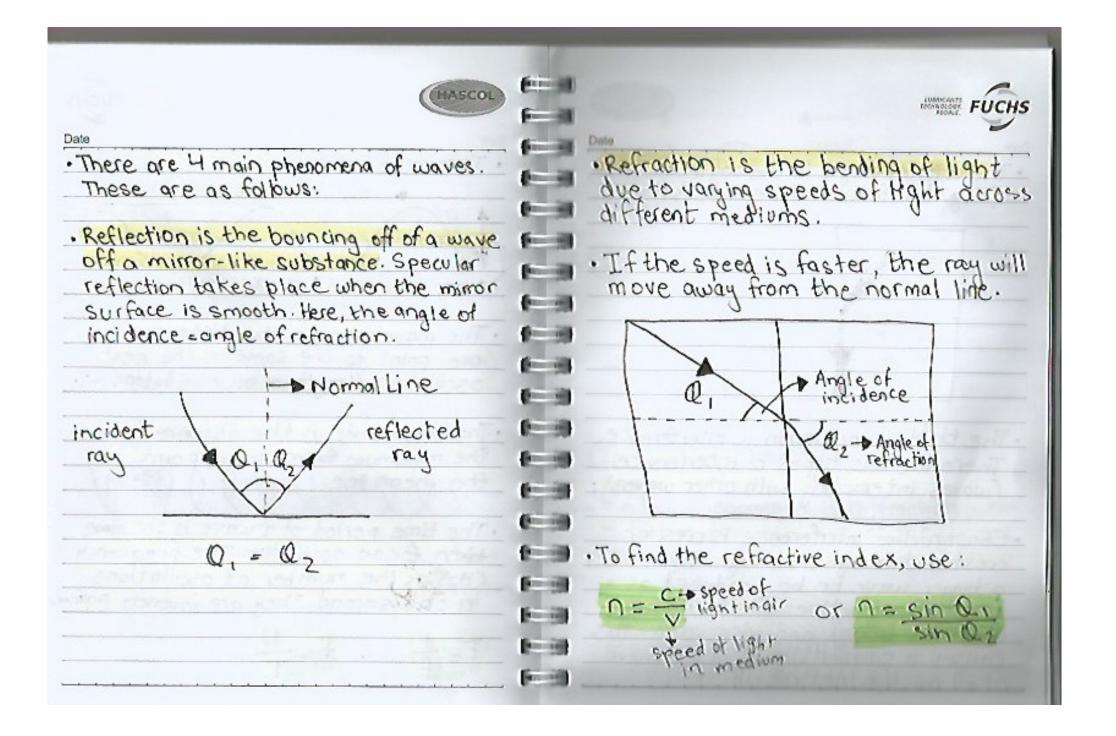
In Physics, density refers to the degree of compactness of a substance.
 Pure water has a density of 2.00.
 Density is given by:
 M - D mass
 P = V - volume.
 Idensity

• To find the relative density of a fluid (relative density is how many more times dense a fluid is compared to water), the following two formulae alcusods

relative density = <u>density</u> of substance density of water

relative density = mass of substance mass of some volume of water . Unlike density, Pressure refers to the constant physical force exerted on an object by something in contact with it. . Measured in Pascals (Pa). 1 Nm" = 1 Pa. . Pressure is defined by: P = f - Force A - Area

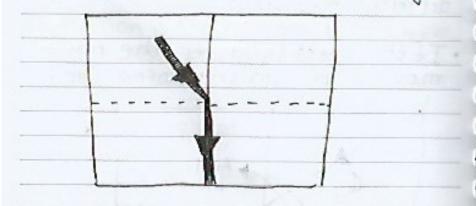




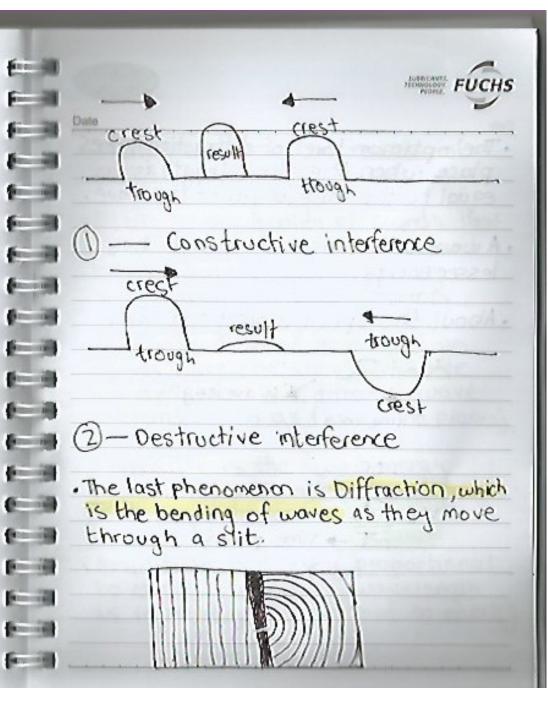


## • The critical angle is the angle at which, if light refracts, it becomes straight.

Date



- The third phenomenon is interference. There are two types of interference (waves interacting with other wavers):
- · Constructive interference is crestion crest and trough on trough. This causes a bigger wave to be produced as a result. Destructive interference is crest on trough and vice vorsa. This causes a smaller wave or no wave at all as the net result.





• The optimum level of diffraction takes place when the slit's length is equal to the wavelength of the wave.

- . A wave with longer wavelength has lesser energy.
- . About the speed/velocity of a wave:

V=fr frequency or V=r T+Time period · There are several <u>characteristics</u> of sound. These are as follows.

HONNESS FUCHS

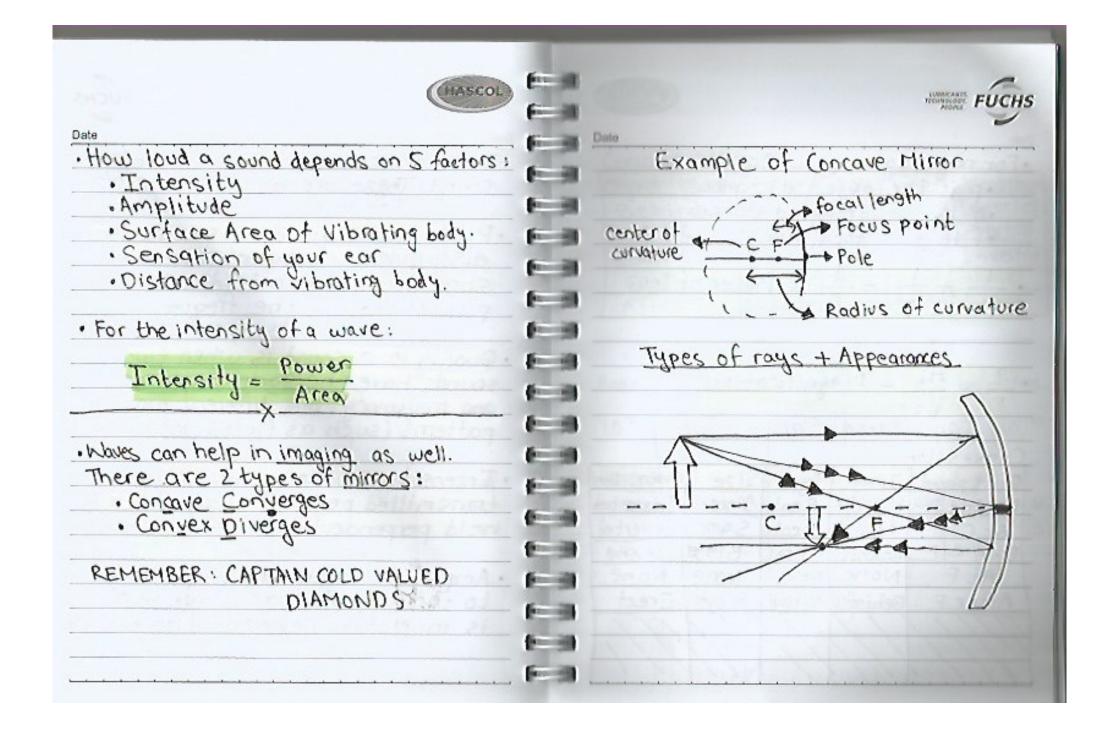
·Pitch is the quality of sounds that distinguishes between grave and shrill sound. Pitch is directly proportional to the frequency.

• Quality in a sound is when two sounds have the same amplitude and frequency, but different wave patterns (such as flutes and prionos).

· Intensity is the sound energy transmitted per unit area, which is held perpendicular.

· Amplitude is directly proportional to energy, whereas wavelength is inversely proportional to energy

Date



HASCOL						LUM RECKI	FL	сня
ate		Date						-
For ray diagrams:		Conces	ve Mirro	C .				
· p is the object distance	Perilling and the second	Object	placed T	magelT	pels	izel	Orion	tation
og is the image distance.	(F-10)	Beh	ind C BI	WACHP	eal S	maller	Inv	ertell
.F is the focal length	6-31	A	tc	Atcir			Inve	
of is the total large	-	Rtwo	CIF		and an other states of the state of the	Bi 994	Inv	ested
· _ + _ = _ = Power of lens	-	At	F	Noner	and the second se	None	and the second division of the second divisio	ne
P q f	6	AFt	CONTRACTOR OF THE OWNER	chind Miller )	virtuoi	Bigge	Ere	ect
	6-10	Tur			1.0			Der a
and the second s	-		Electro	magnetic	Spec	trum	17 4	
2 Hi - Massification	-	0.0.1	Microwave				x-raul	Gammo
$\frac{2}{P} = \frac{H_i}{H_0} = Magnification$	6-3	140010	Ingomave	- miliarco	VISION	101	1031	
1 110	6-3	102	10-2	10-5	6.5×10-	11-8	10-10	10-1
Crave la a	-	10	10	10	0.510	1.		10
Convex long:		TV	Cell phon	Denotes	Helo	heat	Examing	kill
Object placed Image Type Size Orientation	the second se	TV			see	Killer		conce
Behind C Brun C/F Real Smaker Inverter		Coundration	+ Heating	)	Sec	FUSICA		Cell
Atc Atc Real Same Inverted	the second se	-			-	1		cell
Brwn CIF Behind C Real BIGGE Involta	-			-	-			
AtF None None None None	<b>F</b>			1	1			
After F Behindles Virtual Bigger Erect	6-30			-X -				<u></u>
HIAIXIXI NII.	-			- X	-		1019	-
	Real Property lies	0.11	Dan SURST			2.		



ALC: UNK

6 mil

ADD COMP.

ALC: NO

. ....

6.3

And in case

(manual)

## ATOMIC PHYSICS

. The study of radioactive elements and their isotopes.

. The protons and neutrons stick to getter due to a binding force. Apart from the basic parts (protons, neutrons, electrons), they themselves are made up of finier particles called quarks. There are 6 types of quarks.

. UP

Date

- · Down
- . TOP
- · Bottom
  - · Chare . stem

· Rodioactivity is the process of spontaneous disintegration of the nucleus of an atom with the emission of radioactive

rays, including Alpha, Beta and Gamma. . How to show an element: Atomic - A Mass Va -> Symbo! Atomic + Z Number . An element is only radioactive if it atomic mass is greater than 82 amu.

· Three main types of radioactive rays, Alpha (a), Beta (B) and hamma (Y).

· Alpha particles are fast-moving hellum nuclei (2x). They have no electrons and are positive.



· Beta particles are fast-moving electrons (-je), and are negative.

· Gamma rays have no charge, and are photons, or energy particles.

· Characteristics of Radio active Rays:

IDnization power is the number of charge (1) a ray can provide to the medium it flows in.
Alpha: Power of Ionization -(2)
This is because it is 2 electrons short from a full shell.
Beta: Power of Ionization - (1)
They gain one electron from
the medium travelled in.
Gamma: Power of Ionization (2)
Despite Gamma rays having
no charge, they Ionize the
medium to a very Iow level
due to their high speeds.

· Penetration Power (what different rays can pass through and what not: · Alpha: Low (stopped by Paper) · Beta: Medium (stopped by Aluminum) - Gramma: High (stopped by lead concrete) Aluminum Paper Concrete d-High Medium Low · Speeds · Gamma - Speed of light (3×10° »/s) · Beta - Approximately speed of light · Alpha - 10°/0 of speed of light . There are several uses of these radioactive rays.

Date



· Alpha rays are used for detecting the amounts of nutrients in certain crops.

Date

· Beta rays can be used for investigating patients' bodies and tracing organs without surgery.

· mamma rows can be used for sterilising equipment, killing concer cells, detecting breakages and leakages in pipes etc.

. To determine whether a radioactive ray is Alpha, Beta or Gamma, a device called a Geiger Counter is used. • A radioactive isotope (on element with an unstable number of neutrons) a can undergo Radioactive Decay, then become stable. There are Y types of deay:

· Alpha Decay,

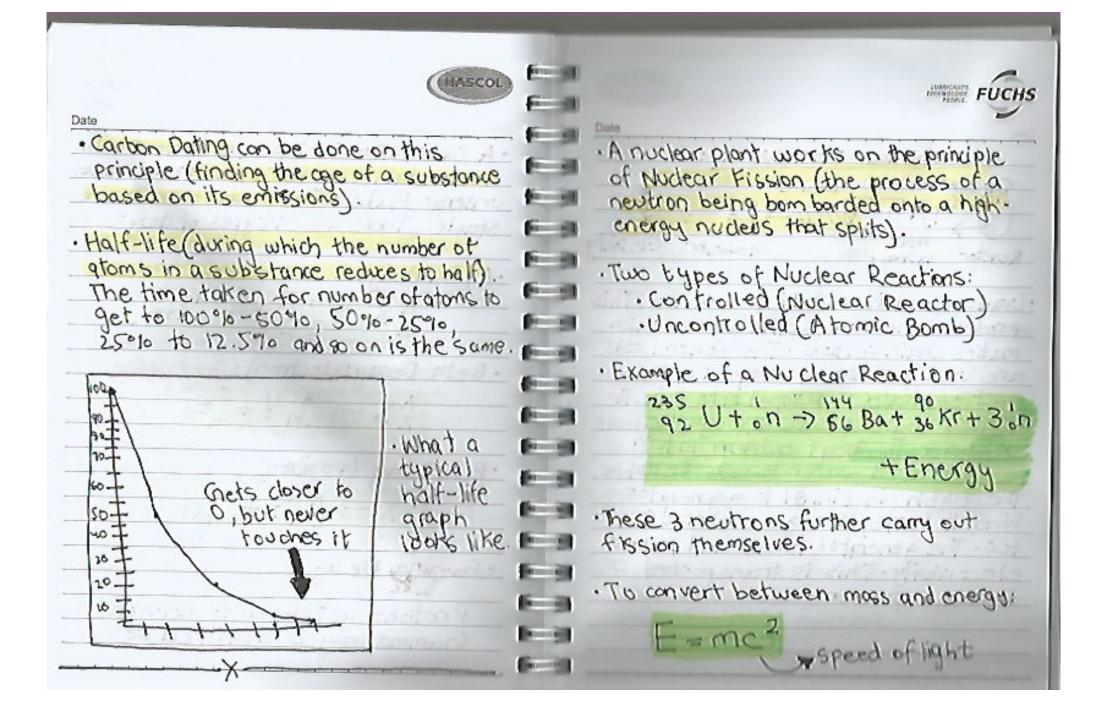
AX-7 2-2 Y+ 2 He

· Beta Decay (electron):

· Beta Decay (positron):  $A \times \rightarrow A \times + e$ 

· hamma Decay :

Protons get, closer to neutrons, Gamma ray is committed boo.





1 m

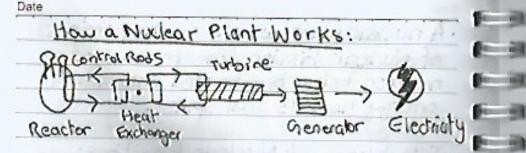
(r = 1

Sec. and

P 1

a second

ALC: NO.



•Uranium fuel is loaded into a reactor. This produces neutrons, splits atoms which makes chain reactions. The control rods are used to control the speed of the reaction. Water is pumped into this. The heated water passes through the heat exchanger. The heat exchanger in turn moves heated water towards the turbine. The turbine is linked to a generator When it turns, the generator does too. The generator turns and produces electricity. This is transported through power grids. ·Apart from Fission, Nuclear Fusion can be used as a power source too. Fusion is the process of two or more. nuclei combining to a create a nucleus of greater mass, Example :. 2H+3H -> Heton - Cando Hission Deuterium Tritium Isotopes of Hydrogen . Since the two isotopes are positive, they repet. Hence, a temperatue of 150 million °C is needed. Fission can be used to carry this out. · Fusion has faults, such as high temperatures, high costs and uncontrollable reactions.



e a

AL. 18.

-

(in 18)

Sec. 120

Married Woman

(h- 10)

P 13

AT 11-

- - - I

6 3

-

4 11

-

and so the second

(HEAT AND THERMODYNAMICS) · Heat is a form of energy that flows from one body to another - specifically from areas of high temperatures to lower temperatures. Its unit is jouk. It is measured by a temperature thermometer (temperature) or a Calorimeter (amount of heat).

 Law of Heat Exchange: Heat lost by hot objects the heat gained by cold object. Thermal equilibrium is reached when heat transfer, as both objects are then of the same temperature (the measure of the hotness foldness of an object). Its units are:
 Kelvin (K) — SI Unit
 Celsius (c) / Centigrade
 Fahrenheit (F) . To convert between K and C:

FUCHS

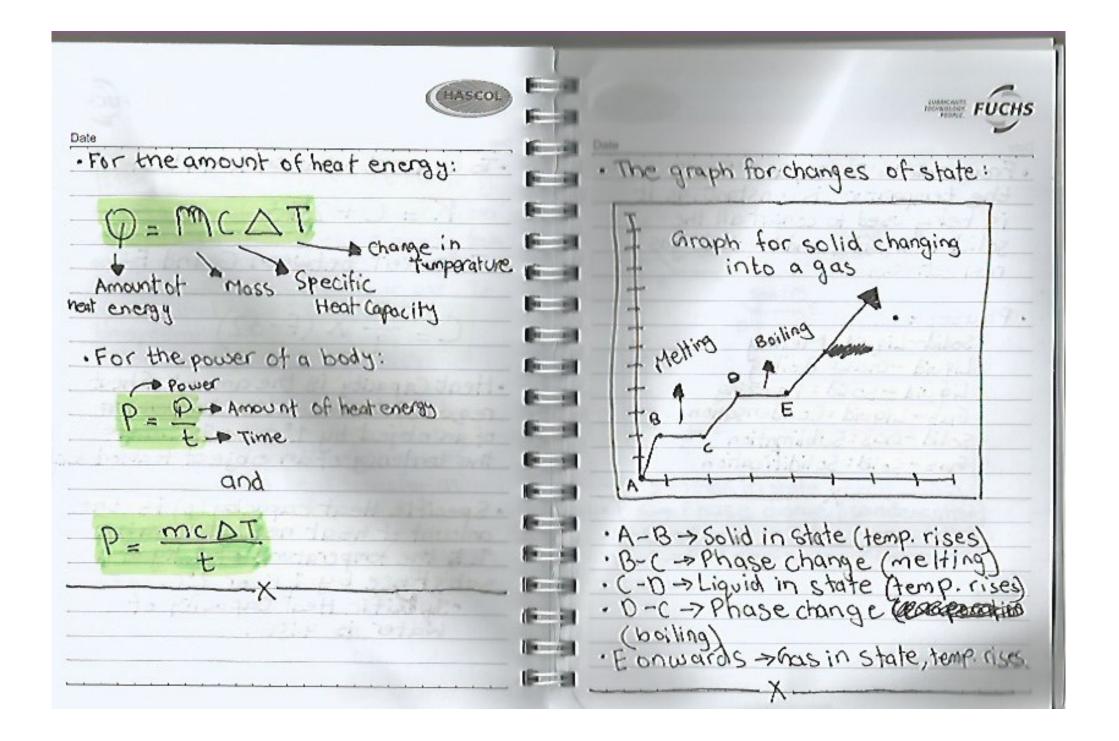
K = C + 273

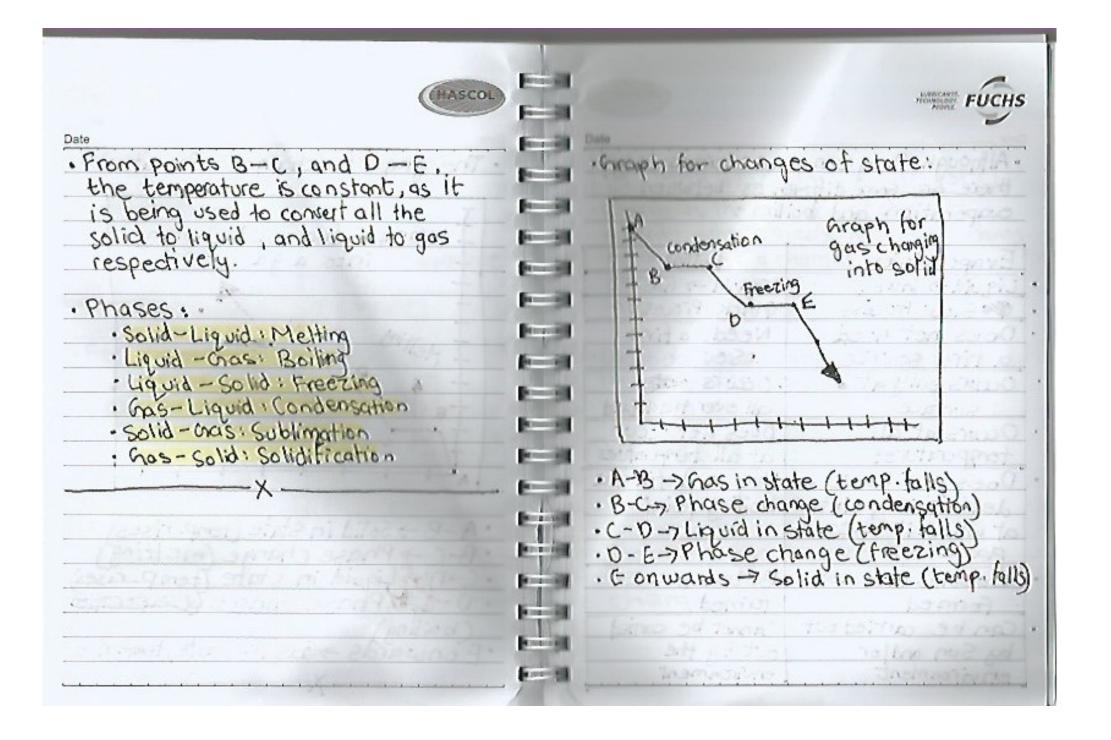
. To convert between C and F:

 $C = \frac{S}{9} \times (F - 32)$ 

·Heat Capacity is the amount of heat required to raise the temperature of an object by 1°C or, 1 K. It is the tendency of an object to absorb heat.

• Specific Heat capacity (c) is the amount of heat needed to raise The the temperature of 1 kg of a substance by 1 K or 1°C. • Specific Heat Capacity-of Water is 4.186.





1000	-	1
1	A STATE	
HAS	OLE	1.15
States of Lot of	1	
	and a state	
No. of Concession, name	and the second se	

ALC: UNK

-

an and the

distantia in

10000

A DOMESTICS

Statute and

Date

· Although they can used interchangeably there are some differences between evaporation and boiling:

Evaporation	Bolling
Liquid-> has	Liquid -7 has
Ge Slow Process	· Quick Process
Does not need .	Needs a fire
a fire source.	Source
Occurs only at .	OCLUIS BOD
surface	all over the liquid
Occurs at all	Does not occur
temperatures	at all temperatures
Does not have a	Has a definitive
definitive point	boiling point .
at which it takes	ai bioris - a-D.
place	1. 9.2.1762-0.
Bubbles are not	Bubbles are
formed	formed.
Can be carried out	Cannot be carried
by Sun and lor	
environment	out by the environment

FUCHS ELECTRICITY AND MAGNETISM · Charge is the tendency of atoms to attract other objects. A charge belongs to an ion, or a charged particle. An atom gets charged by loosing or gaining electrons. · Coulomb's law deals with the internote lar forces of attraction, and the repulsion. between objects. - FX 2, 92 (1)+ Charge Force. The more the force of attraction, the more the product of the charges. 1 1 1 1 1 1



- - - I

a second

1 mil

-

-

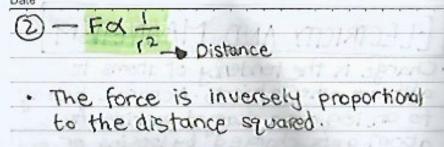
State of the local division in the local div

of Street or other

-

A Description of the local division of the l

11001



3 - F= KQ192 r2

Coulomb's constant is approximately tox 9×10° for air. It is the electric force constant.

· If the force is increased, the charge will increase too.

 It should be noted that Coulomb's constant varies across different mediums. For instance, it is lower for paper. • Static electricity is the charge in stationary (resting) objects. When an object rubs against something, due to heat, the electrons have enough energy to escape the object (abject A, as it can be called). The free electrons stick to another object (we can refer to it as object B). Hence, object A is positive, while B is negative. When they both come into contact, a spark follows, and both objects become neutral.

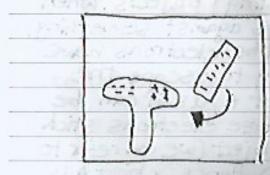
· Electrostatic induction is the phenomenon in which, in a presence of a charged body, the charges on the insulator lother body are distributed in a such a way that like charges come on one side, while unlike charges come on the other end.



-

MARCINE

· Electrostatic induction:



Date

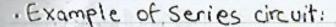
. The word "electrostatic" means having to do with electric charges/fields of stationary objects.

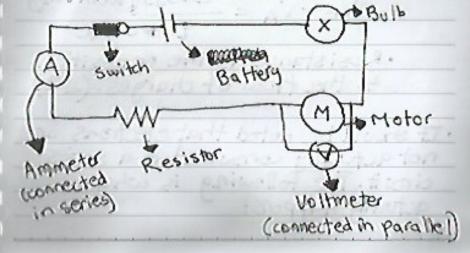
· A circuit is an enclosed path through which electricity flows. On the other hand, current is the rate of flow of negative positive charges. Current is measured by an ammeter in Amperes (A).

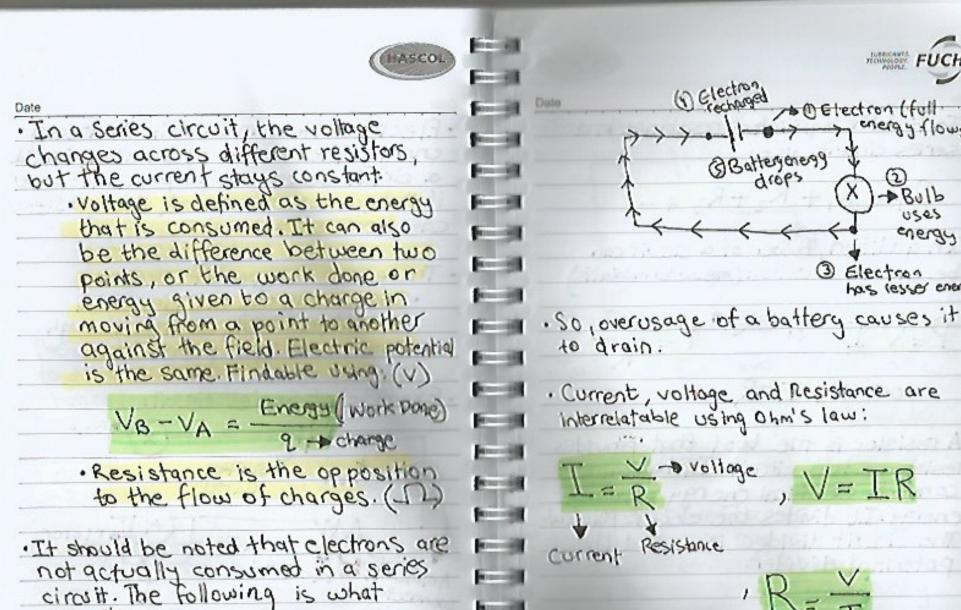
as good -- show

· Electrons flow in a circuit that carry energy. The nucleas of an atom, inside a circuit serves no purpose. However, if it becomes unbalanced, the substance can become radioactive .

- · There are 2 types of circuits : · Series: Only one path in which
  - · Parallel: Multiple paths.







actually happens:

· Current, voltage and Resistance are interrelatable using Ohm's law:

energy flows)

Bulb

has lesser energy

3 Electron

uses

energy



. To find the equivalent resistance in a series circuit, use:

Date

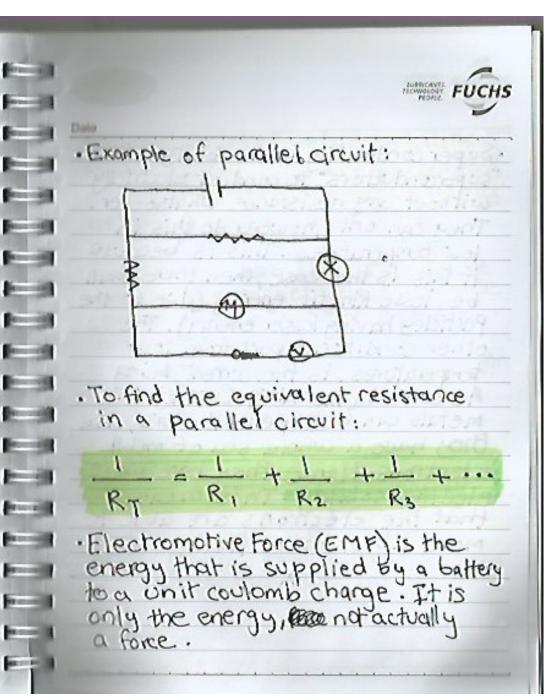
 $R_T = R_1 + R_2 + R_3 + \dots$ 

· In addition, Power of a circuit can be calculated using: (measured in Watts)

P = V T or  $r = P = \frac{V^2}{P}$ 

or P=I2R

•A resistor is the tool that provides resistance to a circuit. It basically converts electrical energy into heat energy. It divides the electric potential. The circuit divider is called the potential divider.



HASCOL

40102-001

**F** 

-

a seal

of the local division of the

at the local division of

-

STATISTICS.

Date

· Super conductivity is the ability of 'superconductors" to conduct electricity without only resistance whatsoever. They can only however do this in low temperatures. This is because if this is the case, then there shall be lesse kinetic energy (due to the particles having lesser energy). The other condition, a part from low temperatures, is the current being AC current (Alternating Current). Only metals can be superconductors, since they have a whole sea of mobile electrons. Hence they have more electronmobility. This means that the electrons are able to move about more freely.

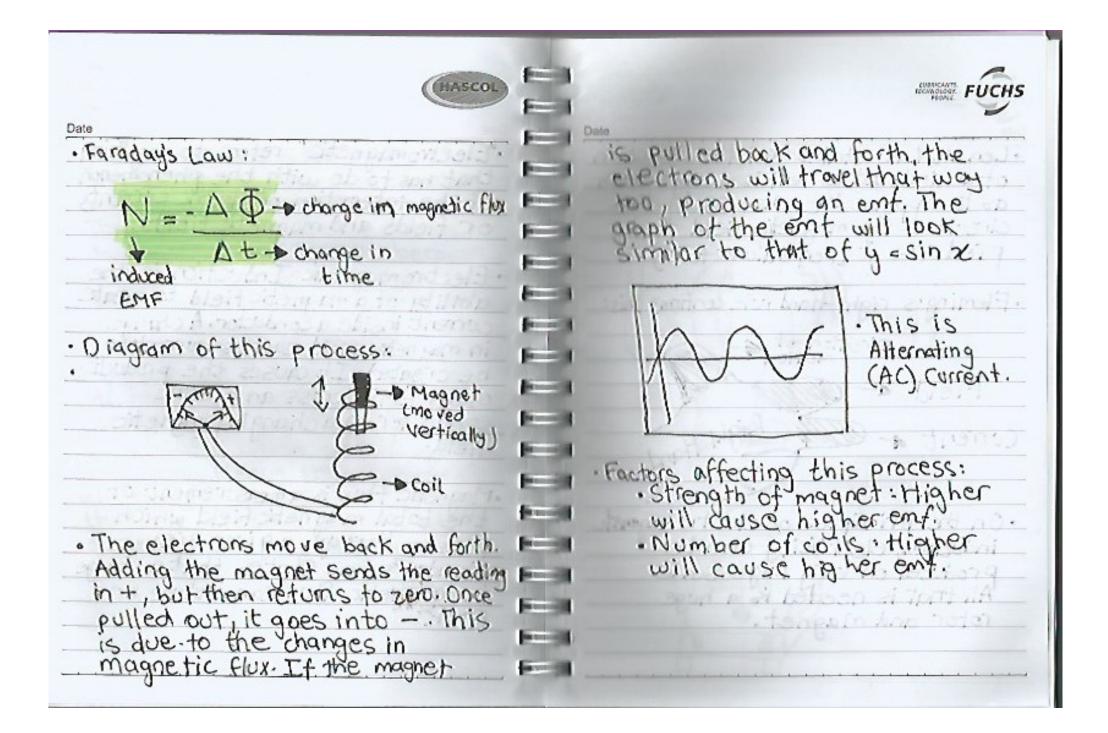
-X -

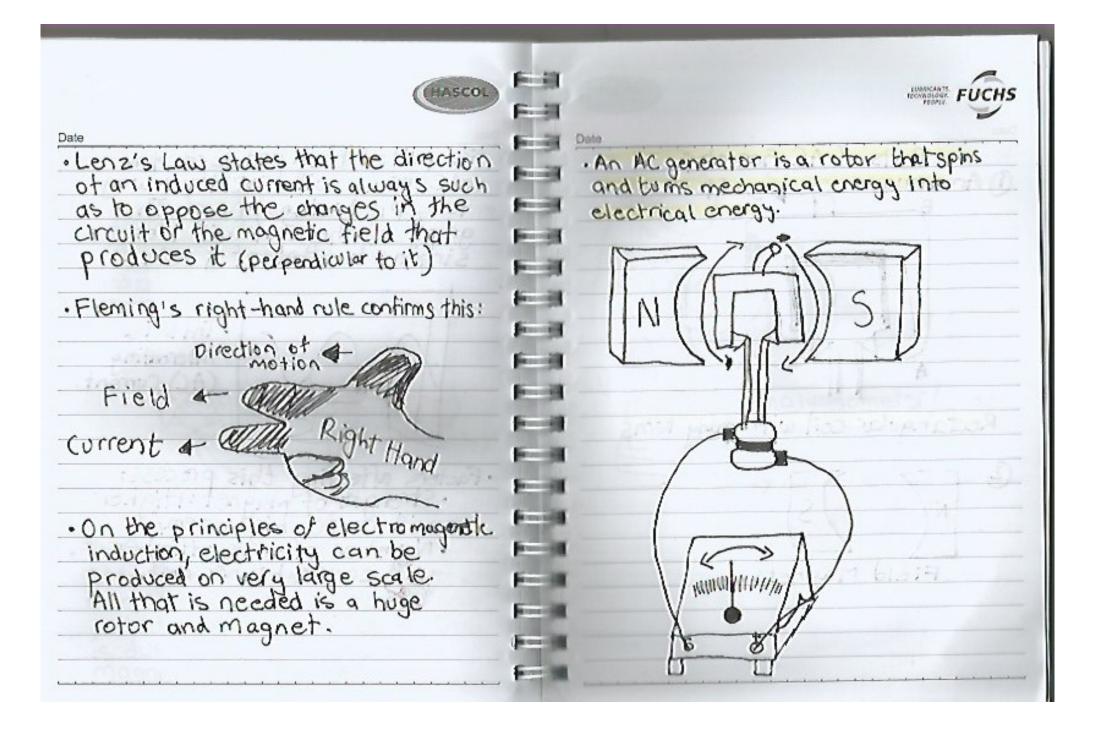
NOVI I'L' 199 MICE

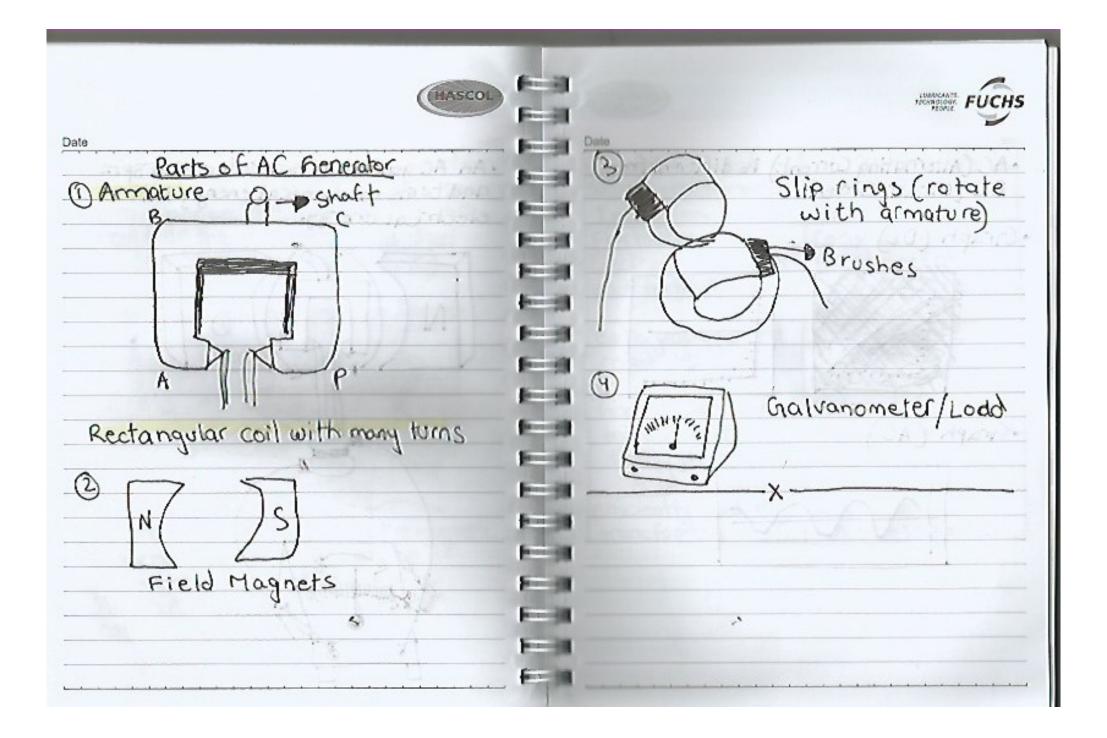
"Electromagnetic" refers to anything. that has to do with the phenomenon of the interaction, of electric currents or fields and magnetic fields.

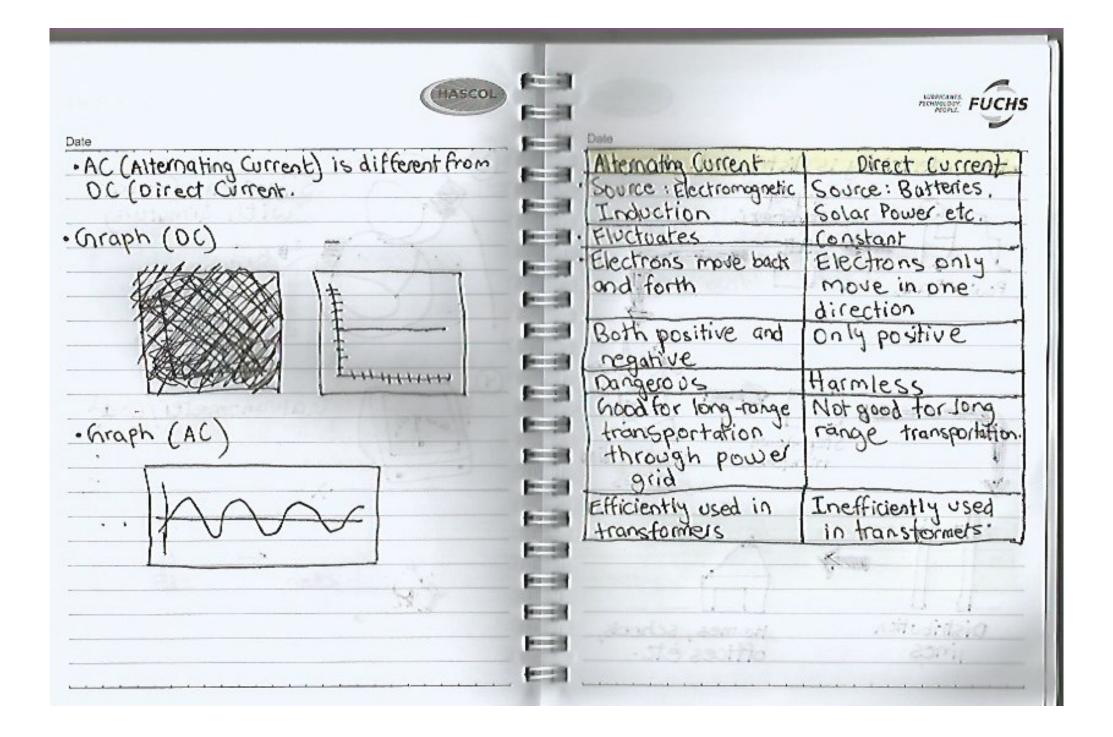
· Electromagnetic Induction is the ability of a magnetic field to create current inside a conductor. A change in magnetic field causes current to be created. It causes the production of an EMF across an electric conductor in a changing manetic field.

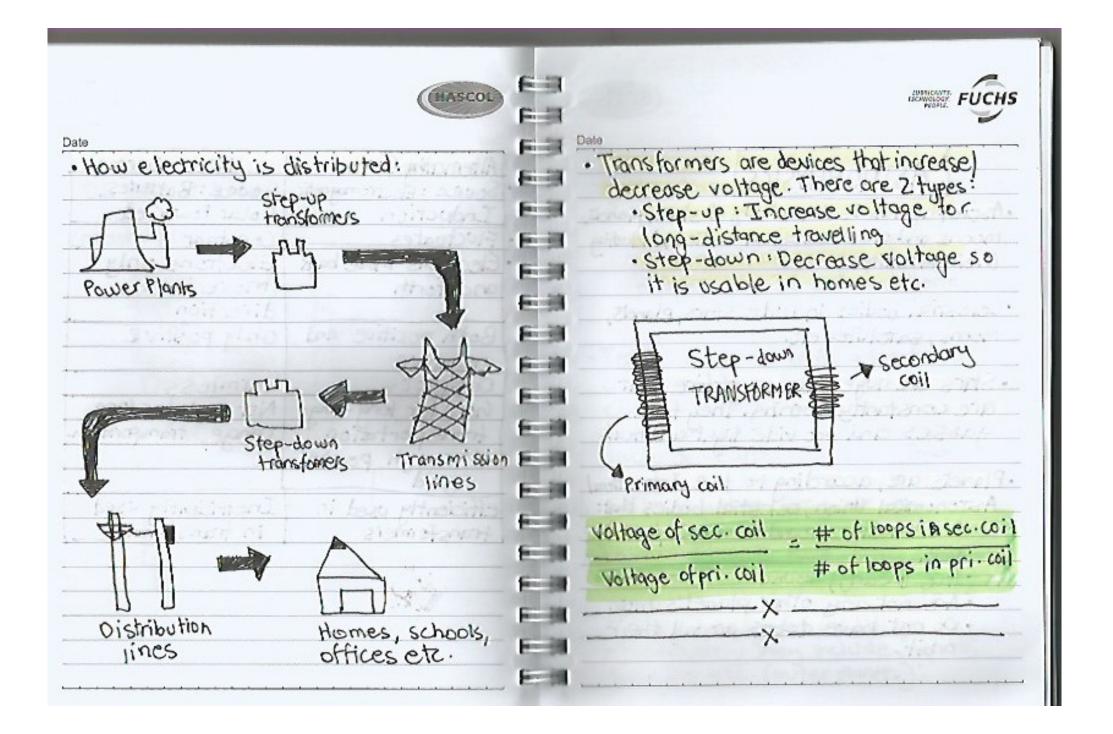
Magnetic flux is a measurement of the total magnetic field which passes through a given area. Faraday's Law States that a change in magnetic flux can induce an EMF.













F

EI

F II

-

F

Street out

and the second

Date

## ASTROPHYSICS

·Astrophysics is the study of stars, planets, moons and other celestial bodies, including their satellites.

·Celestial bodies include stars, planets, moons, satellites etc.

· Stars are huge masses office that are constantly burning. They fuse gasses and provide light and heat.

· Planets are, according to the International Astronomical Union, celestial bodies that: ·Orbit a star around a circular/oval-like Path.

- . Are (nearly) round
- · Are not any other planet's mon
- . Do not have debris around their orbit.

. There are an estimated 10? 4 punets. in the Observable Universe, There are 2 types of planets:

- · Terrestial: Small and Rocky
- · Jovian · hight, made of gasses
- . There are 36 sub-categories of planets.

· Terrestial planets have a smaller size and mass. They have a heavy molten core and topological features like volcances and craters. Examples include Earth Mars etc. They make up the inner part of our Solar System. They are rocky.

· Jovian Planets have larger sizes. They only have small amounts of rocks in cores, and are mostly made of gasses (Hydrogen and Helium in atmosphere). They include Jupiter, Saturn etc. (outer planets).



-

Sec. 1

Sec. 1

-

----

-

States and in case

Date

• All planets revolve around a star and rotate on different axes. Scientists stay that planets continue to rotate in the same manner in which they were when they collided, due to inertia.

• Dwarf Planets are the same as normal planets, except they have debris in their outer surroundings.

· Scientists say planets were formed by the collision and vibration of rocky substances. Due to their great masses, they turned circular.

· A satellite is a moon, planet or star that orbits a planet or star. Moons are generally regarded as "Natural Satellites". . There are many types of satellites : ·Natural Satellites -> Naturally orbit stars or planets · Low Earth Orbits > 160 km away from Earth. Complete an or bit in 90 minutes. Used by military to locate tanks. · Geosynchronous > 24 hours for an orbit. Used for communication at high altitudes. · heostationary relajority of . communication satellites. Orbit once in 24 hours, 14788 km above the Earth. · Sun-synchronous -> 15-16 orbits daily. They, in polar orbits make weather predictions, being fixed relative to the sun. · Satellites work by reflecting Signals back to Earth. An uplink is sent from earth to the satellite,



F-3

-

10 A A

P-il

-

- -

E 1

E

1

F

-

Control of

after which data is processed. Transpondes are used to avoid incoming and outgoing signals. Finally, a downlike is used to send the data from the satellites back to Earth.

· Everything, according to the Big Bang Theory, come into existence about 13.7 billion years ago. It says that at the time there was infinitely dense and Small "dot" that "burst" into existence. It also states that all galaxies are spreading apart from one another. After the dot burst, it was spreading. There was no actual explosion, just the stretching of the universe. That dot to the stretched into gluons, then guarks, and matter was able to "triumph" over antimatter. Then, the Yessential forces were formed (Electromagnetic, Strong & weak Nuclear and Otavity)

Within 10° seconds of the event, ther universe was already a billion km in diameter. Quarks started producing new trons and protons. Within a second, the universe had already spreat over a 100 billion km, and synthesized the first atom - Hydrogen. It was around 10 billion at the time. Within a few minutes, atoms formed stors, galaxies, planets etc.

· Some people question as to what existed before the Big Bang. It is important that we understand that the Big Bang did not just create Space, but Space - Time. Meaning that time did not exist before the Big Bang.

· Scientists "travel back in time" to the Big Bang Using the Redshift Theory. This suggests that as

Date



"galaxies spread out, their wavelong the become larger. Scientists reversed this to "see in the past"

• A telescope is an optical instrument that makes distant objects (celestia) bodies) appear to be nearer. It contains an arrangement of lenses and mirrors, and focuses rays onto one point. Types of telescopes:

- Astrograph : For photographing distant astronomical objects
  Comet See Ker: Searching comets
  Go To : Automatically points to distant objects [celestial bodies.
  - · Infrared: Uses infrared varys

. Etc.

Date

•There are two main categories of teles copes : Refractive and Reflective.

